## TOSS UP

1. EARTH AND SPACE Multiple Choice Which of the following are two optical factors that reduce atmospheric transparency?
W) Reflectivity and absorption
X) Reflectivity and conduction
Y) Absorption and convection
Z) Advection and precipitation

ANSWER: W) REFLECTIVITY AND ABSORPTION

## BONUS

1. EARTH AND SPACE Short Answer Arrange the following four layers of the Earth in order from the Earth's surface to its center: 1) liquid core, 2) lithosphere, 3) asthenosphere, 4) solid core.

ANSWER: 2, 3, 1, 4 (ACCEPT: LITHOSPHERE, ASTHENOSPHERE, LIQUID CORE, SOLID CORE)

## TOSS UP

2. BIOLOGY Multiple Choice A scorpionfish demonstrates which of the following types of coloration?
W) Cryptic
X) Disruptive
Y) Aposematic
Z) Countershading

ANSWER: W) CRYPTIC

## BONUS

2. BIOLOGY Short Answer What is the term for the specialized communal fishing behavior used by humpback whales?

ANSWER: BUBBLE NETTING (BUBBLE NET FEEDING, BUBBLE CLOUDS, LUNGE FEEDING, TRUMPET CALLING)

## TOSS UP

3. MATH Short Answer If $y=-x^{2}$, what is the slope of the tangent line to the graph at $x=4$ ?

ANSWER: -8

## BONUS

3. MATH Short Answer Evaluate $\sum_{k=1}^{30}(3 k+2)$ [the summation from $\boldsymbol{k}$ equals $\mathbf{1}$ to 30 of the quantity 3 k plus 2].

ANSWER: 1455

## TOSS UP

4. PHYSICS Short Answer Given the speed of light in a vacuum to be $3 \times 10^{8}$ meters per second, what is the index of refraction for a medium where the speed of light is found to be 2.5 x $10^{8}$ meters per second?

ANSWER: 1.2

## BONUS

4. PHYSICS Short Answer What is the exact speed in meters per second of an object with a mass of 2 kilograms, executing simple harmonic motion with amplitude of 2 meters, at the end of a spring with a spring constant of 8 newtons per meter, when the displacement of the object is half the maximum amplitude?

ANSWER: $2 \sqrt{3}$
5. ENERGY Multiple Choice Which of the following sectors consumes the most energy in the US?
W) Motor vehicle transportation
$X$ ) Commercial and residential buildings
Y) Air travel
Z) Electronics

ANSWER: X) COMMERCIAL AND RESIDENTIAL BUILDINGS

## BONUS

5. ENERGY Multiple Choice On a per gram basis, why do carbohydrates contain less energy than hydrocarbons?
W) Carbohydrates contain oxygen and hydrocarbons do not
X) Carbohydrates cannot be burned and hydrocarbons can
Y) Hydrocarbons come from fossilized animals and plants but carbohydrates come from live or recently dead plants and animals
Z) Carbohydrates can be eaten but hydrocarbons cannot

ANSWER: W) CARBOHYDRATES CONTAIN OXYGEN AND HYDROCARBONS DO NOT

## TOSS UP

6. CHEMISTRY Multiple Choice Two reactants are mixed in a bomb calorimeter and a reaction occurs. The temperature of the calorimeter rises. Which of the following statements regarding the reaction is true?
W) The reaction is absorbing thermal energy
$X$ ) The reaction is endothermic
$Y$ ) The reaction is releasing thermal energy
$Z$ ) The reaction energy is equal to $q \Delta T$

## ANSWER: Y) THE REACTION IS RELEASING THERMAL ENERGY

## BONUS

6. CHEMISTRY Short Answer Determine the entropy change, including sign, in joules per mole Kelvin when one mole of nitrogen gas is added to hydrogen gas to form ammonia gas. Assume standard molar entropies to be 192 for dinitrogen, 131 for dihydrogen, and 193 for ammonia.

ANSWER: -199

## TOSS UP

7. EARTH AND SPACE Multiple Choice Which of the following is NOT useful for dating extremely old fossils?
W) Potassium-40
X) Carbon-14
Y) Uranium-238
Z) Thorium-232

ANSWER: X) CARBON-14

## BONUS

7. EARTH AND SPACE Short Answer What is the wavelength in kilometers of a shallow-water wave with a speed of 600 kilometers per hour and period of 15 minutes?

ANSWER: 150

## TOSS UP

8. BIOLOGY Multiple Choice The demographic transition model helps explain why which of the following occurs?
W) Birth rates fall before death rates
$X$ ) Death rates fall before birth rates
Y) Development requires large populations
Z) Industrialization begets large populations

ANSWER: X) DEATH RATES FALL BEFORE BIRTH RATES

## BONUS

8. BIOLOGY Multiple Choice Which of these alternative breeding strategies results in an individual indirectly passing on its genes?
W) Young male birds display in a lek with an older, unrelated male who takes all the copulations X) Young female birds sneak an egg into another female's nest to be incubated
Y) Non-breeding helpers are unrelated to offspring of a breeding pair
Z) Non-breeding helpers are full siblings to offspring of a breeding pair

ANSWER: Z) NON-BREEDING HELPERS ARE FULL SIBLINGS TO OFFSPRING OF A BREEDING PAIR

## TOSS UP

9. MATH Multiple Choice If $|x|>-5$ [the absolute value of $x$ greater than negative 5] is graphed on a number line, which of the following could describe its graph?
W) Segment
X) Ray
Y) Line
Z) Two opposite rays, not sharing a common endpoint

ANSWER: Y) LINE

## BONUS

9. MATH Short Answer Find the $x$-coordinates of every point on the graph of $f(x)=x^{3}+3 x^{2}-9 x+3$ where the tangent line is horizontal.

ANSWER: 1 AND -3

## TOSS UP

10. PHYSICS Multiple Choice An object is placed 20 centimeters in front of a positive lens with a focal length of 10 centimeters. Which of the following kinds of image will be produced behind this lens?
W) Upright and virtual
X) Upright and real
Y) Inverted and virtual
Z) Inverted and real

ANSWER: Z) INVERTED AND REAL

## BONUS

10. PHYSICS Short Answer A person weighs 300 pounds on a scale. Given g as 9.8 meters per second squared, find the person's weight in newtons to two significant figures.

ANSWER: 1300

## TOSS UP

11. CHEMISTRY Multiple Choice If equal quantities by mass of gases $\mathrm{A}, \mathrm{B}, \mathrm{C}$, and D are added to a closed container of fixed volume, and the total pressure in the container is 760 torr, the partial pressure of $A$ is which of the following?
W) $25 \%$ of the total pressure
X) Dependent on the size of the container
Y) Directly related to its mole fraction
Z) A function of its molar mass

ANSWER: Y) DIRECTLY RELATED TO ITS MOLE FRACTION

## BONUS

11. CHEMISTRY Short Answer At $25^{\circ}$ Celsius, the standard reduction potential of $\mathrm{Fe}^{3+}$ to $\mathrm{Fe}^{2+}$ is +0.77 volts. Providing your answer to two significant figures, what is the potential, in volts, of a platinum electrode dipped in a solution of 0.0005 molar $\mathrm{Fe}^{3+}$ and 0.05 molar $\mathrm{Fe}^{2+}$ ?

ANSWER: 0.89

## TOSS UP

12. BIOLOGY Short Answer What compound is the final nitrogenous waste product that is excreted by humans?

ANSWER: UREA

## BONUS

12. BIOLOGY Multiple Choice Hundreds of fossils of an extinct quail species are found on a small Pacific island. The fossil femur is longer than that of modern quails from the mainland, and the fossil ulna is shorter. Why might these differences exist between the extinct quail and modern specimens?
W) The extinct quail was adapted to jumping higher because of a higher concentration of predators on islands
$X$ ) The extinct quail was likely evolving flightlessness
Y) The extinct quail fossils were from juveniles
Z) The extinct quail was a hybrid between the mainland species and another, so far undiscovered, island species

ANSWER: X) THE EXTINCT QUAIL WAS LIKELY EVOLVING FLIGHTLESSNESS

## TOSS UP

13. EARTH AND SPACE Multiple Choice Scientific investigations in the Gulf of Mexico have documented a large area of the Louisiana continental shelf with seasonally-depleted oxygen levels. Which of the following is the term for this oxygen depletion?
W) Barotrauma
X) Hypoxia
Y) Hyperbaria
Z) The bends

ANSWER: X) HYPOXIA

## BONUS

13. EARTH AND SPACE Short Answer What is the term for the apparent contradiction between an eternal, infinite universe and the observation that the night sky is dark?

ANSWER: OLBER'S PARADOX

## TOSS UP

14. BIOLOGY Short Answer In the evolutionary tree of life, wings in birds and butterflies are an example of what kind of structures?

ANSWER: HOMOPLASIES (ACCEPT: ANALOGOUS STRUCTURES; HOMOPLASTIC STRUCTURES)

## BONUS

14. BIOLOGY Multiple Choice Why is treatment of carbon monoxide poisoning with an oxygen mask NOT effective?
W) Carbon monoxide binds up oxygen
$X$ ) Carbon monoxide has a higher affinity for hemoglobin than oxygen does
Y ) The pressure at which the oxygen is delivered is approximately $25 \%$ higher than the effective pressure
Z) Carbon monoxide changes the binding site of hemoglobin, making it unsuitable for binding oxygen

ANSWER: X) CARBON MONOXIDE HAS A HIGHER AFFINITY FOR HEMOGLOBIN THAN OXYGEN DOES

## TOSS UP

15. CHEMISTRY Multiple Choice A nimble minded Science Bowl contestant makes a concentrated solution of sodium hydroxide which she adds to a solution of potassium permanganate. The permanganate ion reacts to form which of the following?
W) $\mathrm{MnO}_{4}{ }^{2-}$
X) $\mathrm{MnO}_{4}^{-}$
Y) $\mathrm{Mn}_{2} \mathrm{O}_{3}$
Z) $\mathrm{Mn}^{2+}$

ANSWER: W) $\mathrm{MnO}_{4}{ }^{2-}$

## BONUS

15. CHEMISTRY Short Answer Transition metals can have iodine atoms as ligands; however, the iodine atom is quite large in size. Based on this information, what is the point group of titanium (IV) iodide?

ANSWER: $T_{d}$ (ACCEPT: TETRAHEDRAL)

## TOSS UP

16. MATH Short Answer An acute angle has a measure of $2 a+2 b$. What is the difference between the degree measures of the supplement and complement of that angle?

ANSWER: 90

## BONUS

16. MATH Short Answer What is the sum of the first 50 terms of the arithmetic sequence with second term 35 and fourth term 55 ?

ANSWER: 13,500

## TOSS UP

17. PHYSICS Multiple Choice A small marble is placed at the bottom of a hemispherical bowl. When the marble is displaced a small amount from the center, it oscillates about the bottom of the bowl. Which of the following changes would increase the period of oscillation of the marble?
W) Using a marble of larger mass
X) Using a marble of smaller mass
Y) Using a bowl of larger radius
Z) Using a bowl of smaller radius

ANSWER: Y) USING A BOWL OF LARGER RADIUS

## BONUS

17. PHYSICS Multiple Choice A current of 100 milliamperes flows through an ohmic circuit element across which the potential drop is 5 volts. Which of the following is the resistance in ohms?
W) 50
X) 20
Y) 0.2
Z) 0.05

ANSWER: W) 50

## TOSS UP

18. ENERGY Multiple Choice On average, which of the following consumes the most electricity in an office building?
W) Space heating
X) Air cooling
Y) Lighting
Z) Office equipment

ANSWER: Y) LIGHTING

## BONUS

18. ENERGY Multiple Choice Which of the following is NOT a way of storing energy?
W) Pumping water into a water tower on a hill
X) Dynamic braking on a diesel-electric locomotive
Y) Formation of NADPH in photosynthesis
Z) Storing molten salt

ANSWER: X) DYNAMIC BRAKING ON A DIESEL-ELECTRIC LOCOMOTIVE

## TOSS UP

19. CHEMISTRY Multiple Choice Consider the following reaction:
$\mathrm{Cl}_{2}{ }_{(\mathrm{g})}+2 \mathrm{NaOH}{ }_{(\mathrm{aq})} \rightarrow \mathrm{NaOCl}_{(\mathrm{aq})}+\mathrm{H}_{2} \mathrm{O}_{(\mathrm{l})}+\mathrm{NaCl}_{(\mathrm{aq})}$. Which of the following is true?
W) This is a disproportionation reaction
$X$ ) The oxidation number of the oxygen changes
Y) The oxidation number of the sodium changes
Z) The oxidation number of the chlorine remains constant

ANSWER: W) THIS IS A DISPROPORTIONATION REACTION
$[(G)=$ GAS; $(A Q)=A Q U E O U S ;(L)=$ LIQUID $]$

## BONUS

19. CHEMISTRY Short Answer What is the electron count for $\operatorname{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Br}_{2}$ ?

ANSWER: 16

## TOSS UP

20. EARTH AND SPACE Multiple Choice Which of the following is a metamorphic rock?
W) Serpentinite [suhr-pen-TUH-nyt]
X) Diorite
Y) Syenite [SY-uh-nyt]
Z) Dunite

ANSWER: W) SERPENTINITE

## BONUS

20. EARTH AND SPACE Multiple Choice Which of the following is a Secchi [SEK-ee] disk used to measure?
W) pH
X) Conductivity
Y) Turbidity
Z) Salinity

ANSWER: Y) TURBIDITY

## TOSS UP

21. BIOLOGY Multiple Choice Which of the following organisms does NOT undergo alternation of generations in its life cycle?
W) Polytrichum moss [pol-ee-TRIK-uhm]
X) Diatom
Y) Laminaria brown algae
Z) Fern

ANSWER: X) DIATOM

## BONUS

21. BIOLOGY Multiple Choice In a countercurrent exchange system, arteries and veins are located near one another, allowing the transfer of arterial heat energy to the cooler venous blood. Conservation of heat energy through this system decreases the need for additional energy to warm the chilled venous blood. Which of the following best illustrates the greatest need for heat conservation using a countercurrent blood flow?
W) Stomachs of beluga whales
X) Ears of Asian elephants
Y) Flippers of Atlantic dolphins
Z) Breast muscles of Canadian geese

ANSWER: Y) FLIPPERS OF ATLANTIC DOLPHINS

TOSS UP
22. MATH Short Answer No three points of a particular set of 4 points are collinear. How many lines are determined by pairs of those points?

ANSWER: 6

## BONUS

22. MATH Short Answer If $\sin \theta+\cos \theta=\frac{1}{3}$ [sine of theta plus cosine of theta equals one third], find $\sin 2 \theta$.

ANSWER: $-\frac{8}{9}$

## TOSS UP

23. PHYSICS Multiple Choice A torsional pendulum consists of a disk suspended from a torsional spring. The torsional pendulum is set into oscillation when the disk is displaced through a small angle. Which of the following changes would increase the frequency oscillation of the torsional pendulum?
W) Decrease the moment of inertia of the disk
X) Increase the moment of inertia of the disk
Y) Decrease the torsional spring constant of the spring
Z) Decrease the angle through which the disk was displaced

## ANSWER: W) DECREASE THE MOMENT OF INERTIA OF THE DISK

## BONUS

23. PHYSICS Short Answer An object floats with 10\% of its volume below the surface of the water in a pond. Find the average density of the object in kilograms per cubic meter.

ANSWER: 100

## TOSS UP

24. ENERGY Multiple Choice About how many tons of coal is required to produce the same amount of electricity produced by a typical power plant in one year?
W) 250,000
X) 500,000
Y) 1 million
Z) 4 million

ANSWER: Z) 4 MILLION

## BONUS

24. ENERGY Short Answer By number, rank the following four choices in order of increasing potential to impact climate change in the next 20 years: 1) CFC-12, 2) methane, 3) carbon dioxide, 4) nitrous oxide.

ANSWER: 3, 2, 4, 1

## TOSS UP

25. CHEMISTRY Multiple Choice A DoE researcher runs a redox reaction for a Daniell cell: $\mathrm{Cu}^{2+}{ }_{(\text {aq) }}+\mathrm{Zn}{ }_{(\mathrm{s})} \rightarrow \mathrm{Zn}^{2+}{ }_{(\mathrm{aq})}+\mathrm{Cu}_{\text {(s) }}$. Given that the equilibrium constant for this reaction is 1.65 $\times 10^{37}$ at 25 degrees Celsius, the $\mathrm{E}^{0}$ [ $E$ naught] and the $\Delta \mathrm{G}^{0}$ [delta $\mathbf{G}$ zero] for this reaction are which of the following?
W) Both positive
X) Both negative
Y) Both zero
Z) $E^{0}$ is positive and $\Delta G^{0}$ is negative

ANSWER: Z) $E^{0}$ IS POSITIVE AND $\Delta G^{0}$ IS NEGATIVE
$[(A Q)=$ AQUEOUS; $(S)=$ SOLID $]$

## BONUS

25. CHEMISTRY Multiple Choice In free radical polymerization [pol-lim-er-uh-ZAY-shuhn], many different substituents can be employed to stabilize the chain end radical. All except which of the following are strong stabilizers of the radical?
W) CN
X) $\mathrm{C}_{6} \mathrm{H}_{5}$
Y) $\mathrm{OCH}_{3}$
Z) $\mathrm{CO}_{2} \mathrm{CH}_{3}$

ANSWER: Y) $\mathrm{OCH}_{3}$

