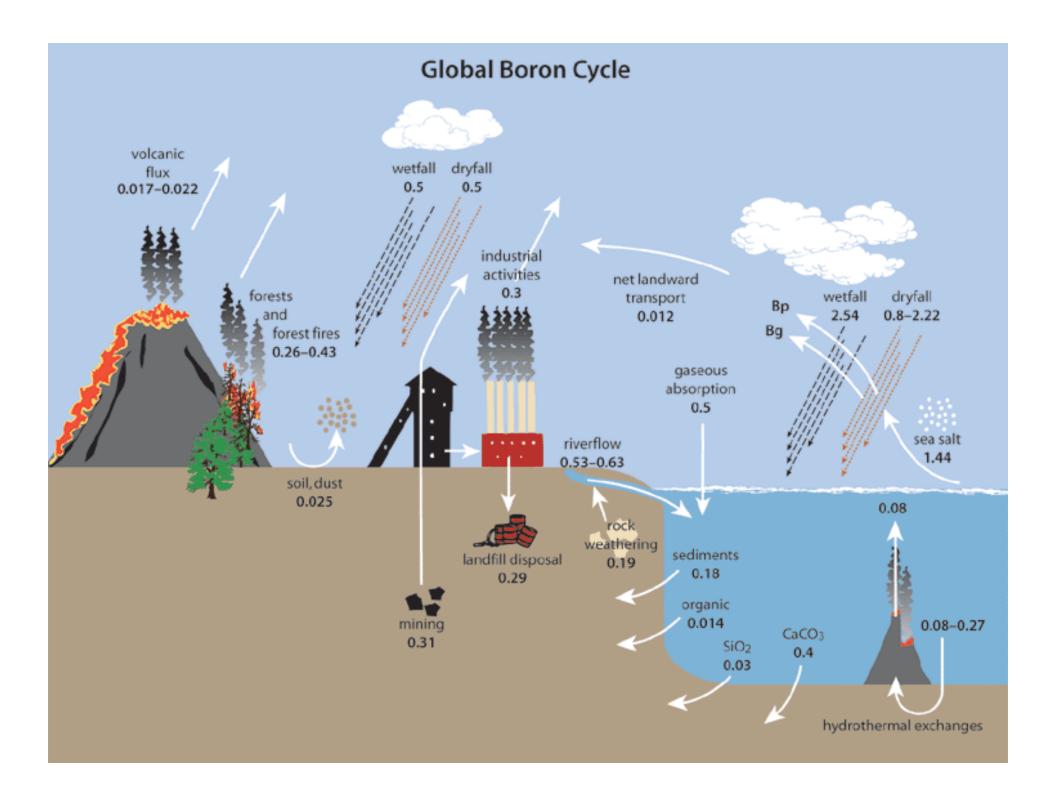
Chasing N Atoms: The Global Nitrogen Cycle



William H. Schlesinger, President

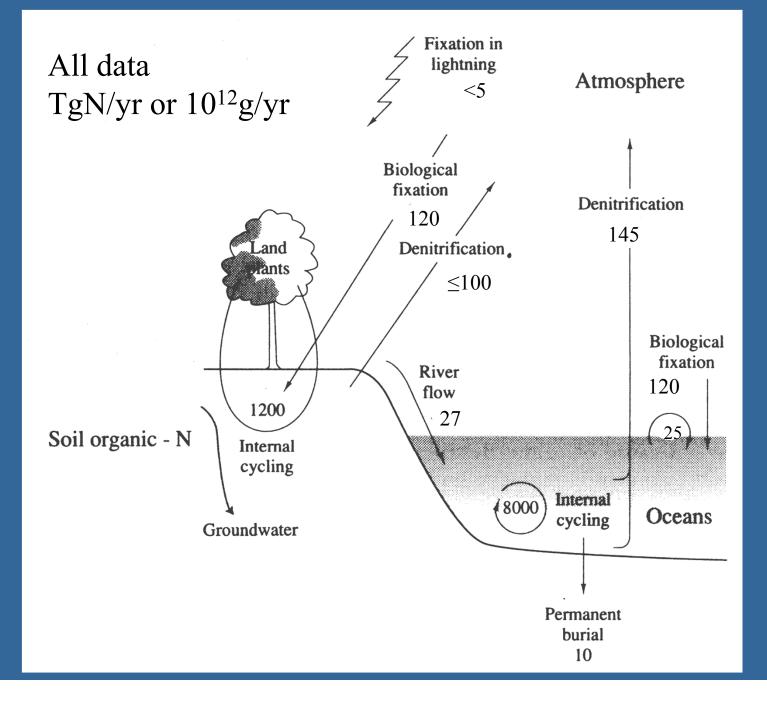




Human Perturbation of Global Cycles

Element	Juvenile Flux	Chemical Weathering	Natural Cycle	Biospheric Recycling Ratio	Human Mobilization	Human Enhancement
	(1)	(2)	(3)	3/(1+2)	(4)	4/(1+2)
В	0.02	0.19	8.8	42	0.58	2.8
С	30	210	107,000	446	8,700	36.3
Ν	0.1	20	9,200	458	221	11.0
Р	~0	2	1,000	500	25	12.5
S	10	70	450	5.6	130	1.6
Cl	2	260	120	0.46	170	0.65
Ca	120	500	2,300	3.7	65	0.10
Fe	6	1.5	40	5.3	840	112.0
Cu	0.05	0.056	2.5	23.6	11	104.0
Hg	0.0005	0.0002	0.003	4.3	0.0023	3.3

The Global Nitrogen Cycle - Pre-Industrial





Growers Special[™] 12-6-6

GUARANTEED ANALYSIS

Todal 81

1.86% Nitrate Nitrogen 1.28% Ammoniacal Nitrogen 8.86% Urea Nitrogen* Available Phosphate (P ₂ O ₅) Soluble Potash (K ₂ O)	
Boron (B) 0.02% Copper (Cu) 0.05% Iron (Fe) 0.25% Total Manganese (Mn) 0.05% 0.05% Soluble Manganese 0.05% Zinc (Zn) 0.05%	0%

Derived From Primary Plant Nutrient Sources: Nitrate of Potash, Ammoniated Phosphate, Urea Formaldehyde. Secondary Plant Nutrient Sources Derived From: Sodium Borate, Copper Sulphate, Iron Sulphate, Manganese Sulphate, Zinc Sulphate. *7.34% Slowly Available Nitrogen From Urea Formaldehyde.

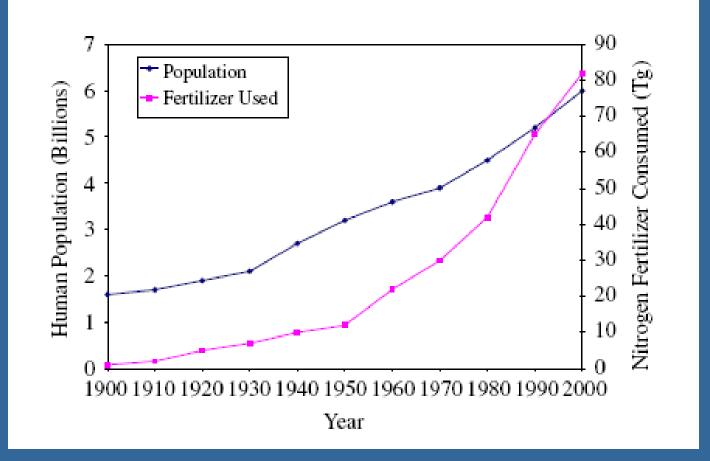
DIRECTIONS FOR USE

370-2305

Hi-Yield® Growers Special™ is designed for controlled feeding of container plants, trees, shrubs, and lawns. The slow release formula in Growers Special™ reduces the risk of nitrogen burn and is chlorine free.

CONTAINER PLANTS: Apply one teaspoonful per 6 inch pot and 2 tablesoons per square foot of soil surface in large containers avon 6 works





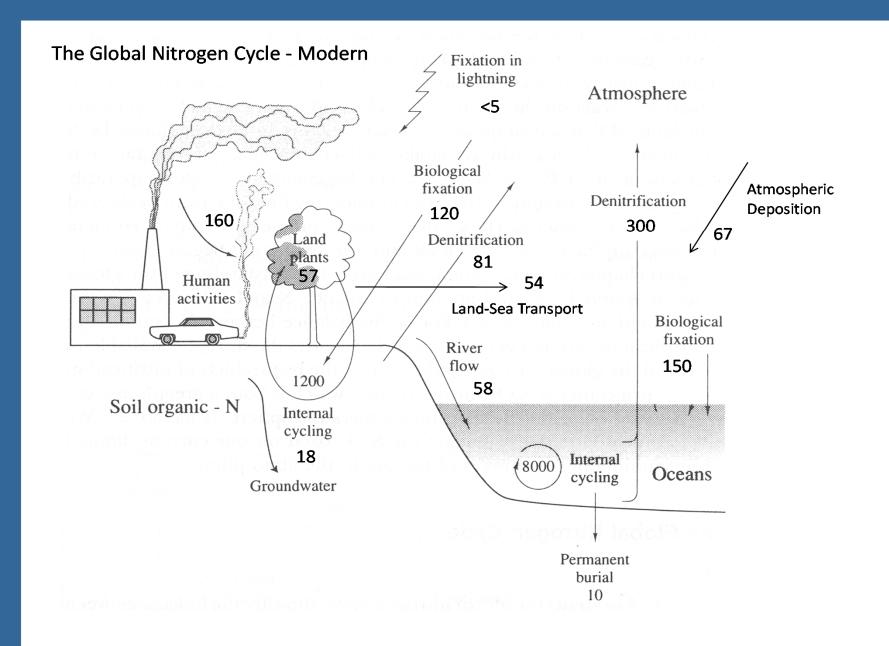
Population increase and use of nitrogen fertilizer from 1900 to 2000.

Aneja et al. 2008.

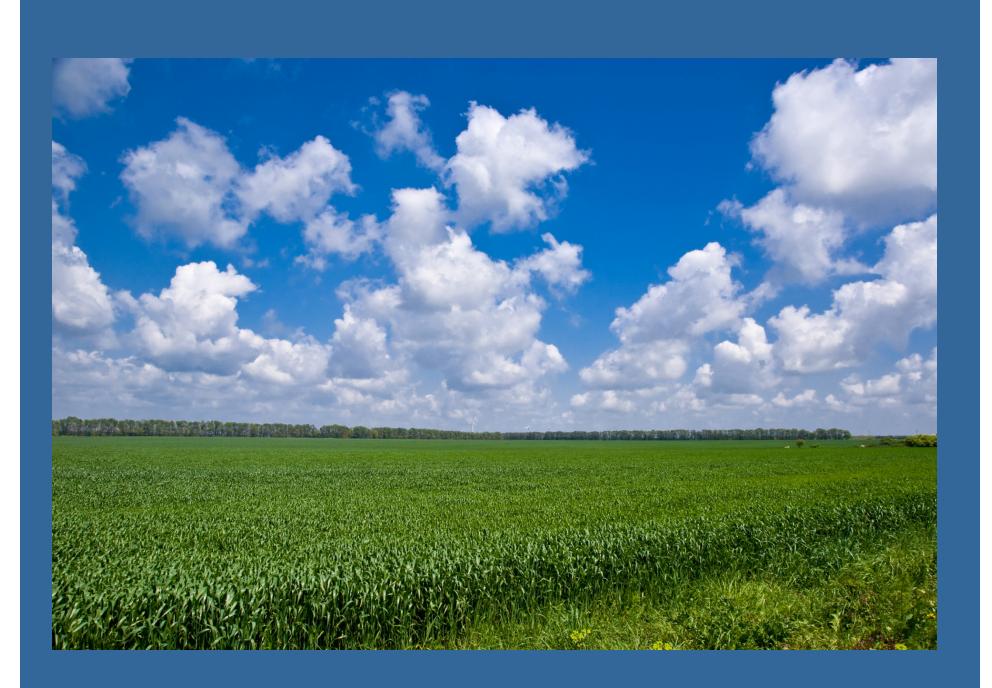
Vaclav Smil

Enriching the Earth

Fritz Haber, Carl Bosch, and the Transformation of World Food Production



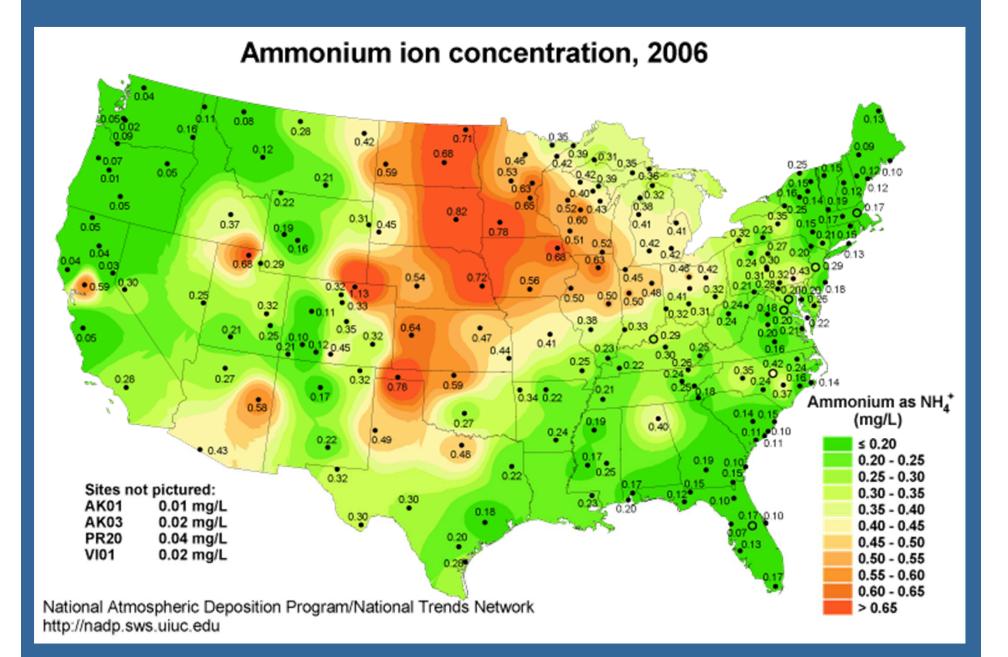
Where does this nitrogen go?

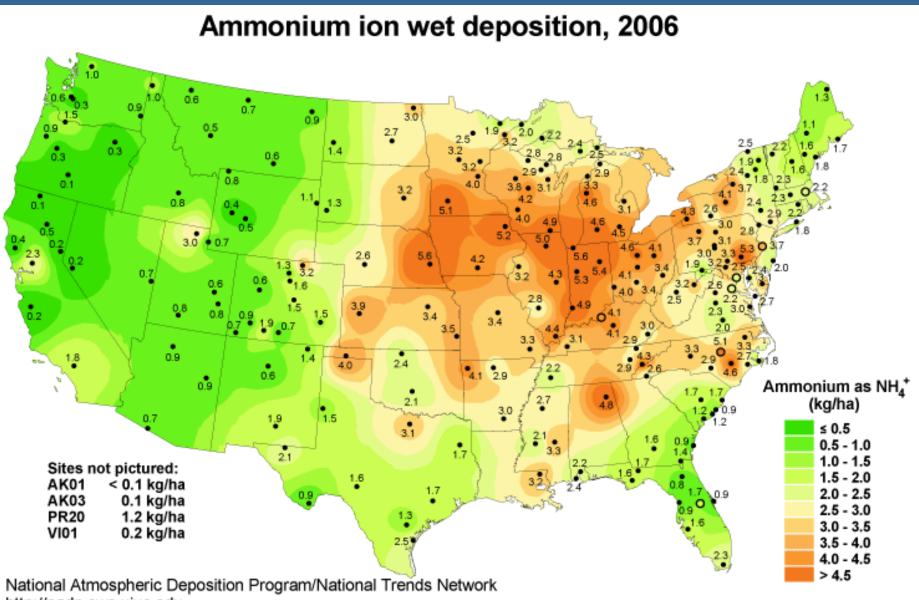


Form of N	Labeled N recovered				
	kg ha⁻¹	percentage of applied, %			
		67 kg N ha ⁻¹			
Plant	20.7	31			
Soil					
Inorganic	2.8	4			
Organic	24.3	36			
Total	47.8	71			
N unaccounted for	19.2	29			
		134 kg N ha ⁻¹			
Plant	45.2	34			
Soil					
Inorganic	3.9	3			
Organic	39.8	30			
Total	88.9	66			
N unaccounted for	45.1	34			
		201 kg N ha ⁻¹			
Plant	74.6	37			
Soil					
Inorganic	13.2	7			
Organic	41.8	21			
Total	129.6	64			
N unaccounted for	71.4	36			
		268 kg N ha ⁻¹			
Plant	94.9	35			
Soil					
Inorganic	30.7	11			
Organic	48.6	18			
Total	174.2	65			
N unaccounted for	93.8	35			

Table 6. Mass balance of ¹⁵N-labeled fertilizer N as affected by long-term N application rate in continuous corn production. Values are 3-yr means of data collected annually at Monmouth, IL, from 1994 to 1996.

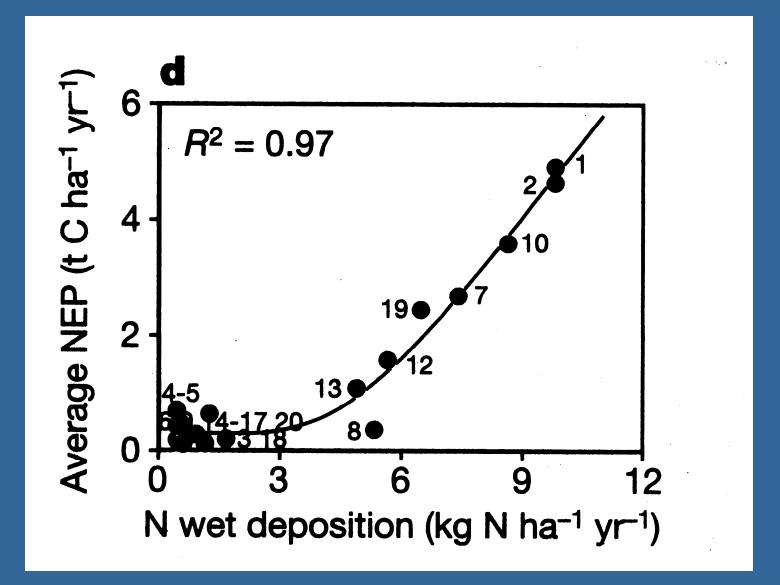
Stevens et al. 2005.





http://nadp.sws.uiuc.edu



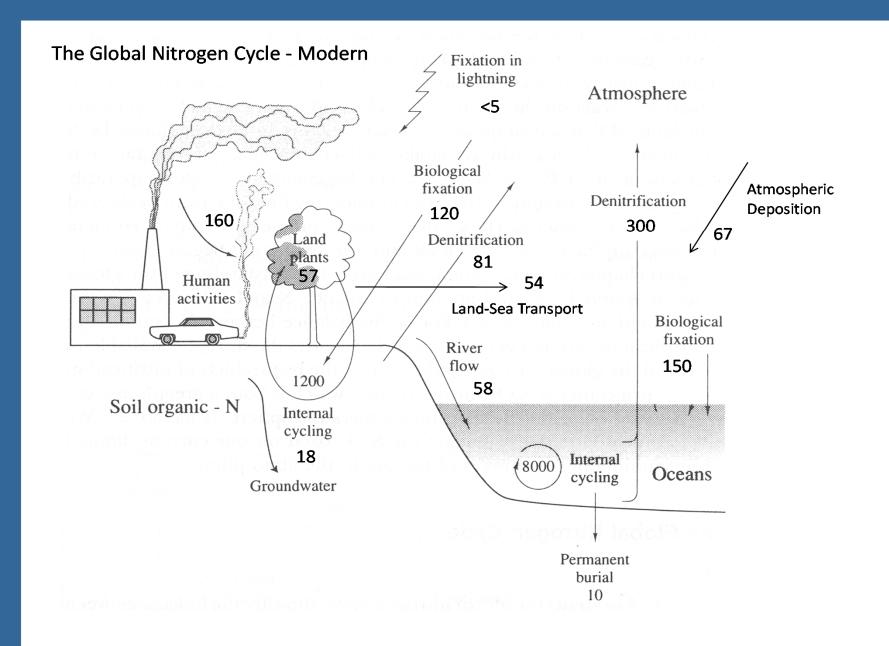


From: Magnani et al. 2007

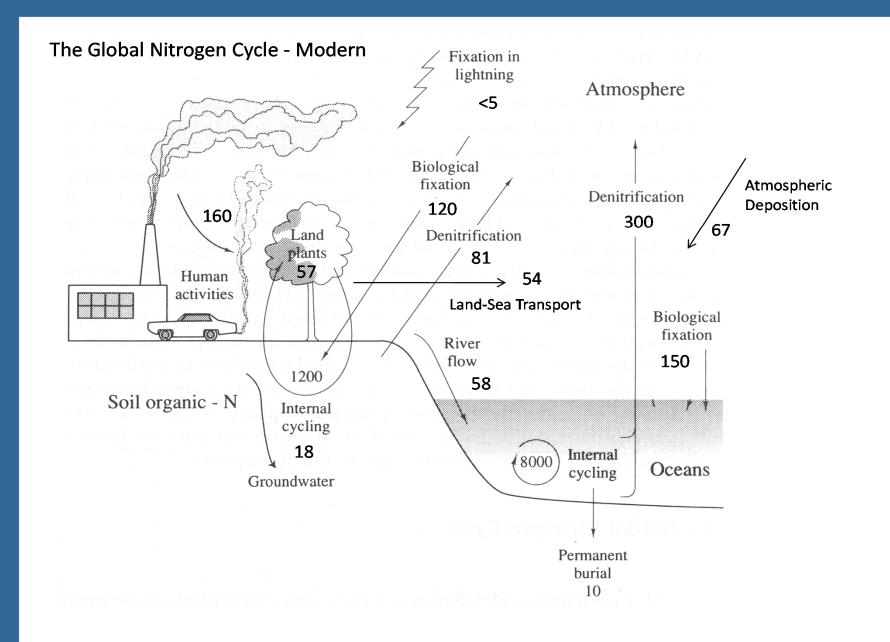
Table 1.

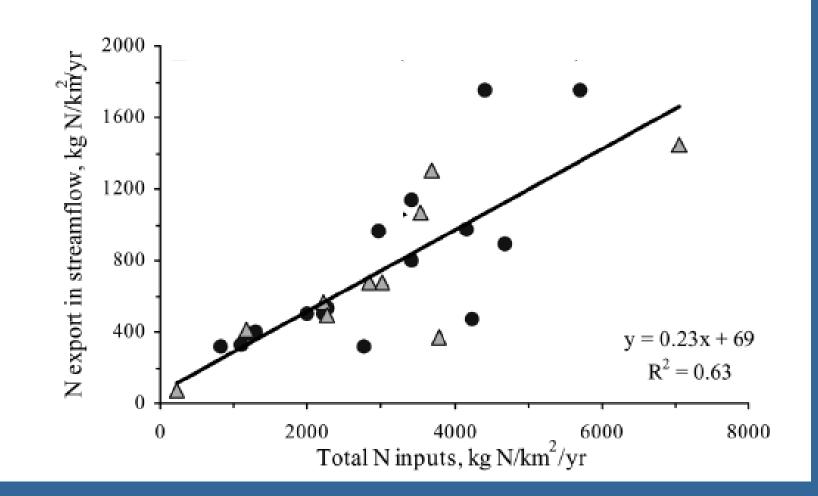
Ecosystem	Age (years)	Method	Total application (kg N/ha)		Percent recovery in				Total	Reference		
type					Plants Litter	Litter	Soil		Leachate	Gaseous	measured	
							Inorganic	Organic		flux	recovery (%)	
Pinus resinosa	50	(Treatement)-	276	6	21		1		1	Tr.	23	MaGill et al. (1997)
		(Control)	826	6	8		2		15	Tr.	25	
Mixed deciduous	50		276	6	20		1		2	Tr.	23	
			826	6	13		1		Tr.	Tr.	14	
Pinus contorta	11	¹⁵ NH ₄	100	8	17	4	0	41			62	Preston and Mead (1994)
		¹⁵ NO ₃	100	8	16	3	0	38			57	
Pinus elliottii	11	(Treatment)-	56	2	25	9		21			55	Mead and Pritchett (1975)
		(Control)	224		27	6		12			45	
Pinus radiata	16	(Treatment)– (Control)	922	9	15	5		21			50	Neilsen et al. (1992)
Pseudotsuga menziesii	35	¹⁵ NH ₄	5	2	33	22		24	2		81	Koopmans et al. (1996)
			50	2	29	15		22	33		99	
Pinus sylvestris	45		5	2	10	46		20	10		86	
			50	2	17	21		16	17		71	· · · · ·

Schlesinger and Andrews. 2000. Biogeochemistry 48:7-20.

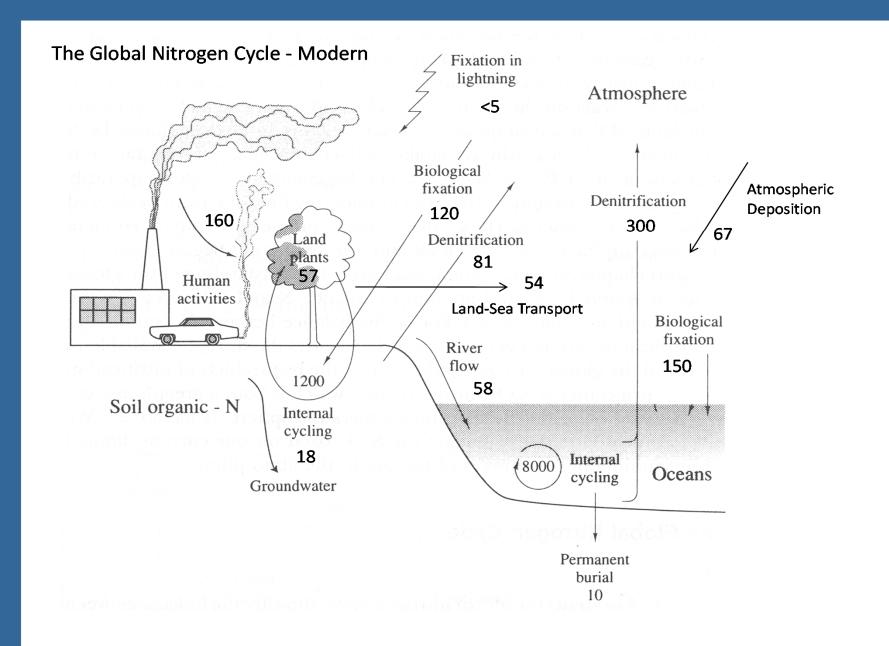


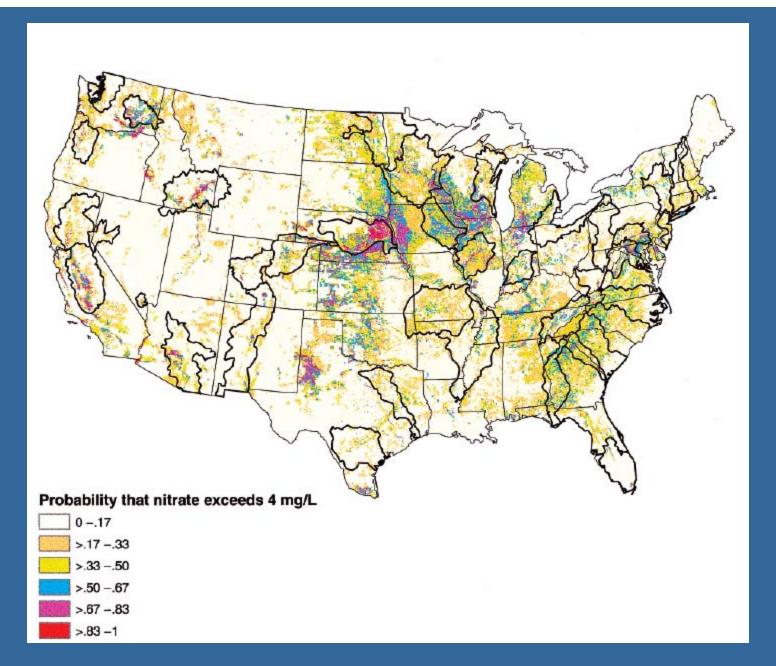






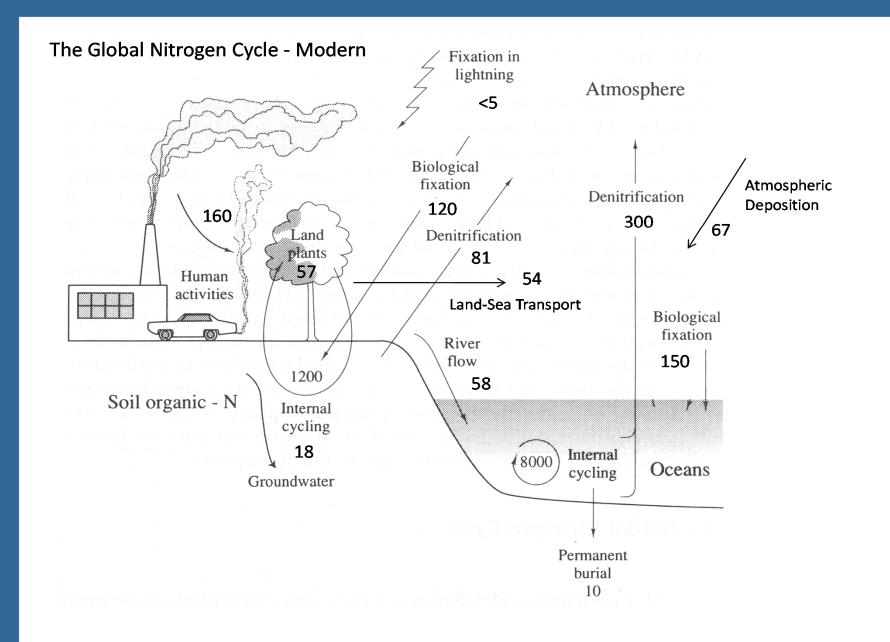
From Van Breeman et al. 2002





Nolan et al. 2002

Calculation of Groundwater Flux $12,666 \text{ km}^3/\text{yr} \text{ X} \quad 10^{12} \text{ l/km}^3 \text{ X} \quad 0.148 = 1,874 \text{ km}^3/\text{yr}$ $1,874 \text{ km}^3/\text{yr} \text{ X} \quad 1.9 \text{ mg/l} = 3.6 \text{ TgN/yr}$ in N. America 3.6 TgN/yr / 0.20 = 18 TgN/yr globally



Denitrification

$5CH_2O + 4H^+ + 4NO_3^- \rightarrow$

 $2N_2 + 5CO_2 + 7H_20$

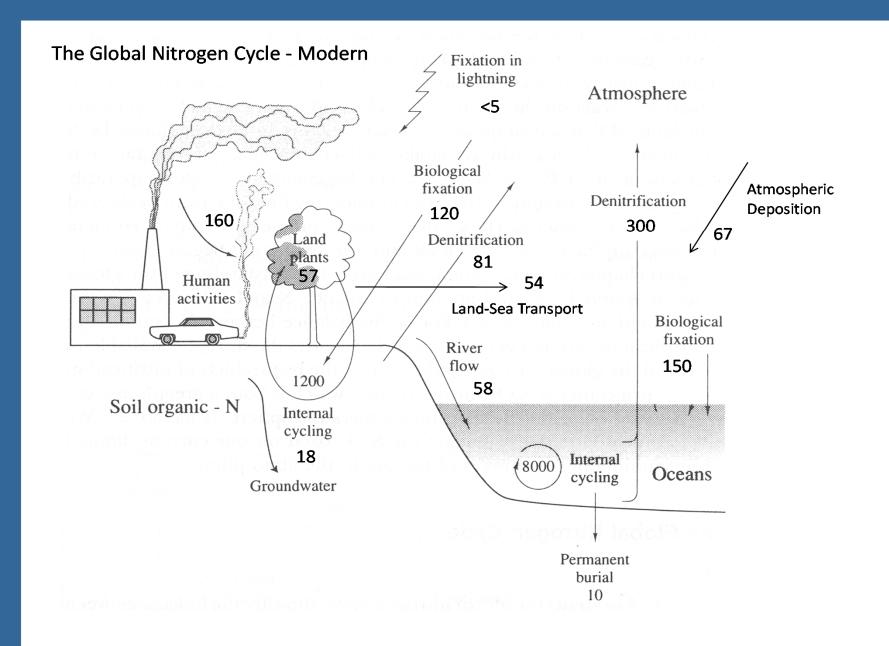
Have you thanked a wetland today?



TABLE 1. Denitrification of land-based N sources based on spatially distributed estimates.

System	Denitrification (Tg N/yr)
Terrestrial Soils	124 (65–175)†
Freshwater Groundwater Lakes and reservoirs Rivers Subtotal	44 (>0–138)‡ 31 (19–43) 35 (20–35) 110 (39–216)
Marine Estuaries Continental shelves Oxygen minimum zones Subtotal	8 (3–10) 46 (>0–70)§ 25 (>0–30?)¶ 79 (3–145)

From: Seitzinger et al. 2006

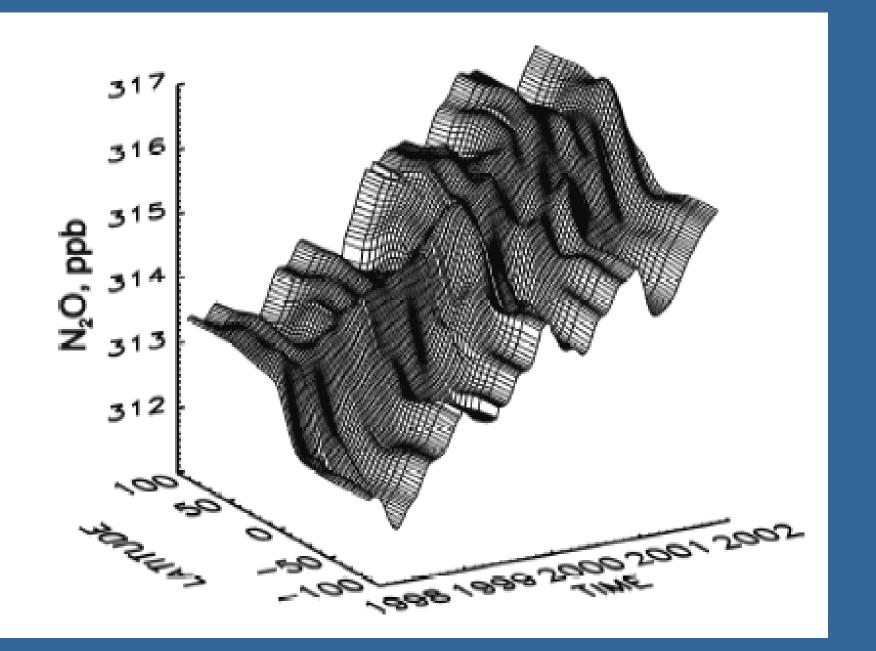


Denitrification

$5CH_2O + 4H^+ + 4NO_3^- \rightarrow$

 $2N_2 + 5CO_2 + 7H_20$

Intermediates include NO and N₂O



Hirsch et al. 2006

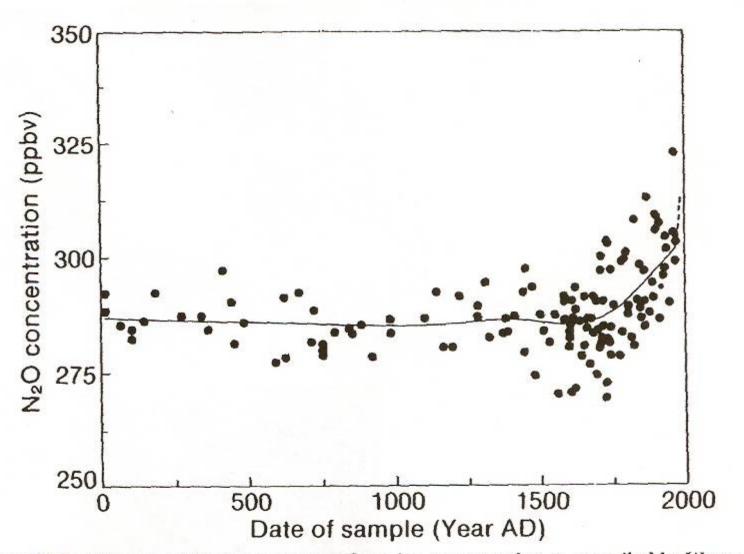


Figure 12.5 Nitrous oxide measurements from ice-core samples, as compiled by Watson et al. (1990).

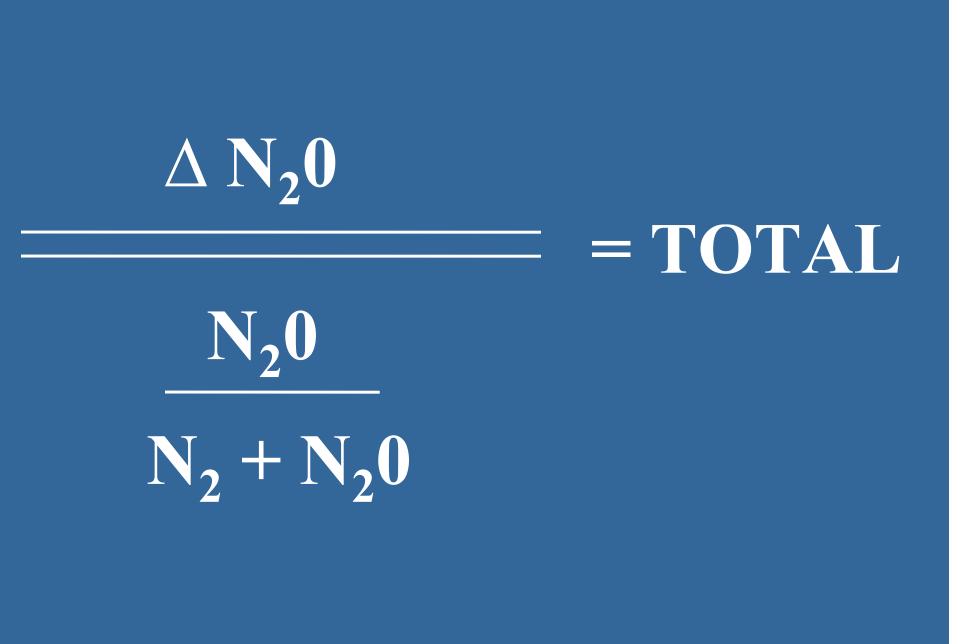


Table 2. Mean N₂O-yield values from various laboratory and field studies of denitrification

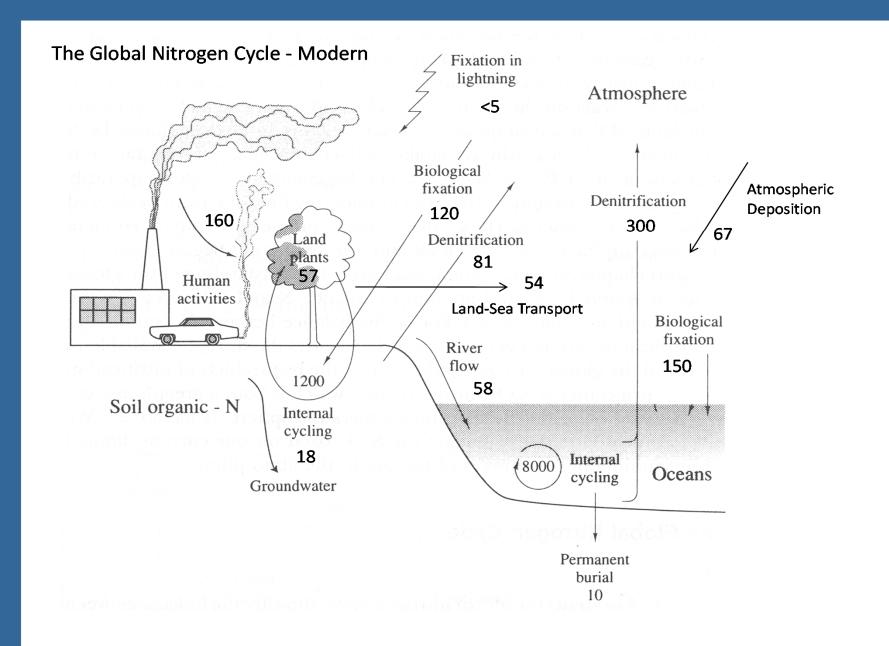
Ecosystem	$N_2O-N/(N_2+N_2O)N$
Agricultural soils	0.375 ± 0.035 (SE)
Soils under natural or recovering vegetation	0.492 ± 0.066 (SE)
Freshwater wetlands and flooded soils	0.082 ± 0.024 (SE)

Full dataset is available as Table S1.

Schlesinger 2009

Calculation of change in denitrification from N_2O

124 Tg (0.37) + 110 Tg (0.082) = 234 Tg (0.246) 4 TgN₂O/yr / 0.25 = 17 TgN/yr



Mass-balance for nitrogen on the Earth's land surface. All values are in TgN/yr (=10¹² g N/yr)

Inputs	Preindustrial	Human derived	Total
Biological N fixation	60 ^a	60 ^b	120
Lightning	5	0	5
Rock weathering	20 ^c	0	20
Industrial N fixation	0	136 ^d	136
Fossil fuel combustion	0	25	25
Total	85	221	306
Fates			
Biospheric increment	0	9	9
Soil accumulation	0	48	48
Riverflow	27	31	58
Groundwater	0	18	18
Denitrification	27 ^e	17	44
Pyrodenitrification	25 ^f	12	37
Atmospheric land-sea transport ^g	6	48	54
Total	85	183	268



